

Lithium-Oxygen Batteries - A Comprehensive Finite Element Model

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Introduction and Background

The Needs for Li-O₂ Batteries:

- Li-O₂ batteries possess 6-10 times greater energy densities compared to Li-ion batteries.
- Great for electric vehicles (EVs) or plug-in hybrid vehicles (PEHVs)
- Many energy potential energy storage uses.

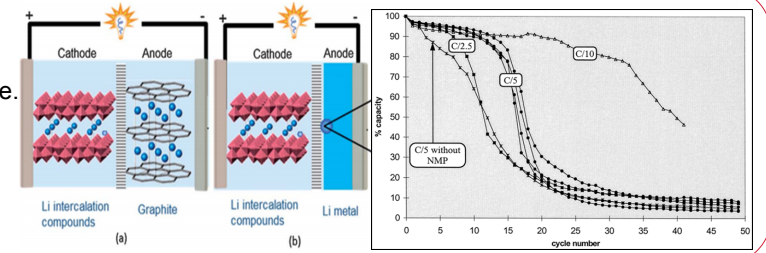


Motivations:

- Dendrite structures grow on anode surfaces.
- Precipitate growth in cathode leads to passivation.
- Identify determinant factors affecting battery performance.

Characteristics of Lithium-Oxygen Batteries:

- Utilizes a lithium metal anode.
- Carbon nanotube cathode structure.
- Oxygen enters and exits the cathode through an oxygen permeable membrane as needed.



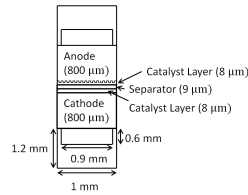
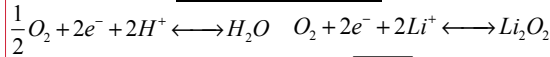
Methods and Results

Fuel Cell Module → Lithium-Oxygen Model

Anode Reactions



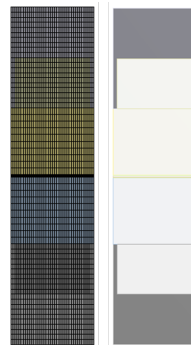
Cathode Reactions



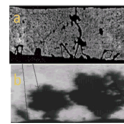
- Construct models of Li₂O₂ batteries with Fuel Cell module within ANSYS Fluent.
- Create models with dendrite growths of different lengths on anode surface.

A Comprehensive Finite Element Model

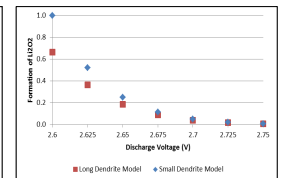
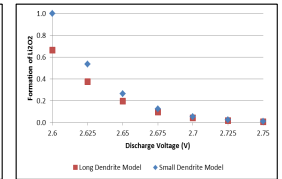
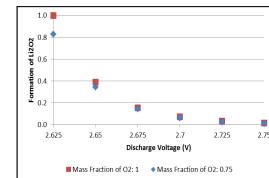
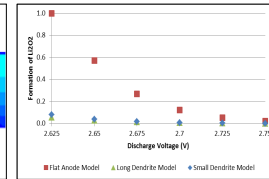
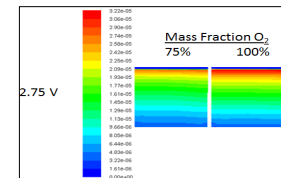
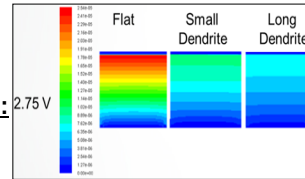
- Simulated discharge in theoretical models to observe several battery parameters such as anode homogeneity, mass fraction of oxygen, mass flow rate of oxygen, and cathode porosity, and their effects on battery discharge.
- Theoretical battery cells discharged at different voltages and their characteristics and performance were compared.
- Total of 150 battery finite element models tested from 2.75 V to 2.6 V.



Anode Homogeneity:



Mass Fraction of O₂:



Discussions and Conclusion

- Found the anode homogeneity affected the discharge parameters of our lithium-oxygen battery greater than porosity or mass fraction of O₂.
- Cathode parameters (mass flow rate of O₂, mass flow rate, porosity) affected the performance of the battery cell to a lesser degree compared to the anode homogeneity.
- Lithium dendrite growth continues to pose a significant problem in the commercial development of lithium-oxygen batteries.
- Model can be further developed to incorporate differing reaction rates, different dendrite structures, dendrite spacing, and cell cycling (charging and discharging).